Name:



1. A quantity of gas is housed in an adjustable container. Its pressure *P* in kPa, temperature *T* in Kelvin, and volume *V* in liters satisfy

$$PV = 8T$$
.

Suppose at time t = 0 (measured in seconds) that the system that the volume and the temperature of the gas are changing and:

$$V = 20\ell \tag{1}$$

$$T = 300K \tag{2}$$

$$\frac{dV}{dt} = 0.1 \frac{\ell}{s} \tag{3}$$

$$\frac{dT}{dt} = 0.2 \frac{K}{s}. (4)$$

1. What is the pressure of the gas at time t = 0?

$$P = \frac{87}{V} = \frac{8.300}{20} = 120 \text{ kPa}$$

2. Use the chain rule to compute dP/dt at time t = 0.

$$\frac{dP}{dt} = \frac{\partial P}{\partial V} \cdot \frac{dV}{dt} + \frac{\partial P}{\partial T} \cdot \frac{dT}{dt}$$

$$= -\frac{8T}{V^2} \cdot \frac{dV}{dt} + \frac{8}{V} \cdot \frac{dT}{dt}$$

$$= -\frac{8.300}{(20)^2} \cdot \frac{1}{10} + \frac{8}{20} \cdot \frac{2}{10}$$

$$= -\frac{6}{10} + \frac{8}{100} = -\frac{52}{100} = -\frac{26}{50} = -\frac{13}{25}$$

2. Suppose a temperature field T(x, y) satisfies $\nabla T = \langle y - 4, x + 2y \rangle$. A bug follows a path $\mathbf{r}(t) = \langle -t, t^2 \rangle$. At what times t does the bug report that d/dt $T(\mathbf{r}(t)) = 0$?

We want
$$\overrightarrow{\nabla} T \cdot \overrightarrow{r}' = 0$$

At time t,
$$\overrightarrow{PT} = \langle Y-4, x+2x \rangle / x=-k, y=k^2$$

$$=46^{3}-36^{2}+4$$

(My bad; misprost

led to cubic;
full credit
for the

Numerically, just one root of